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solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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the following elements in the space provided Modern
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number 3 38. carbon, atomic number 6 39 Modern
chemistry chapter 7 test answer key. neon, atomic
number 10 PART VI Write the ...*

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in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.*

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~~Chapter Test A, continued _____~~ 13. To calculate the number of atoms present in 2.0 mol of an element, you would a. add Avogadro's number of atoms per mole to 2.0 mole. b. subtract Avogadro's number of atoms per mole from 2.0 mole. c.

~~Modern Chemistry Chapter Atoms Test Answers~~
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Assessment Chapter Test A Chapter: The Periodic Law

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Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers...

~~Chapter 6 Modern Chemistry Test Answers~~

PROBLEMS Write the answers to the questions on the line to the left, and show your work in the space provided.

25. 26. MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this isotope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST

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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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experiences, and problem solving. 7 I usually finish at
four in the morning, (2).*

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~~*Modern Chemistry Chapter 7 Test B Answers*~~

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Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon,*

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atomic number 10 PART VI Write the answers to the questions on the line to the left, and show your work in the space provided.

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PROBLEMS Write the answers to the questions on the line to the left, and show your work in the space provided.

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