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Diagrams Made Easy: How to Draw Lewis Dot
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Explain how to use molecular shapes to predict

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molecular polarity. To predict molecule polarity from the shapes, you must first see if the molecule has exactly two atoms. If so, subtract the electronegatives to decide if it's polar. If the molecule has unshared electron pairs on the center atom (bent, trigonal pyramidal), the molecule is polar. If the molecule is linear, trigonal planar, or tetrahedral, it is nonpolar.

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Laboratory 11: Molecular Compounds and Lewis Structures Building 3D Models Use the ball and stick kits provided in class to build 3D models of the molecules after you have drawn the Lewis structures. The balls are color coded as shown in Table 2.

Models Of Molecular Compounds Lab Answers

Lab Models of Molecular Compounds

Name: _____ Introduction. Why should people care about the shapes of molecules? Consider

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that the properties of molecules, including their role in nature, depend primarily on molecular structures. Molecular shape determines a compound's boiling point, freezing point, viscosity, and the amount and type of its ...

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Chemical Bonds, Molecular Models, and Molecular Shapes. BACKGROUND

INFORMATION. The properties of chemical

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compounds are directly related to the ways in which atoms are bonded together to make molecules. Simple “ball-and-stick” models are commonly used to construct representations of some common molecules. These “ball-and-stick” models serve as a three-dimensional representation of an abstract idea.

**Chemical Bonds, Molecular Models, and
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**Models Lab - Chemistry Chemistry 152L,
Molecular Models Lab page 1 Revised 11/8/2009
Molecular Models Lab Objectives 1 Learn about
the structures of covalent compounds and
polyatomic ions 2 Draw Lewis structures based
on valence electrons and the octet rule 3
Construct 3-**

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Chemistry 152L, Molecular Models Lab page 1

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Revised 11/8/2009 Molecular Models Lab
Objectives 1. Learn about the structures of covalent compounds and polyatomic ions. 2. Draw Lewis structures based on valence electrons and the octet rule. 3. Construct 3-dimensional models of molecules and ions with single, double, and triple bonds. 4.

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Alexa Butera Due: January 28, 2020 Period 8/9

Models of Covalent Compounds Purpose: The goal of this lab is to determine the polarity and

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shape of bonds in a molecule. Procedure: 1. Create the dot diagram corresponding to the molecule 2. With this, decide the shape of the molecule 3.

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enzymes are, explore how they work, and learn
why they're needed for your cells';
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To learn how to draw spatial representations and Newman projections, molecular models are useful. These give a simple representation of the geometry of the molecules. Atoms are represented by different colored balls, and bonds are represented by sticks or tubes. Molecular models make the differentiation of different isomers and conformers much easier.

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Learn about the structures of covalent
compounds and polyatomic ions. 2. Draw Lewis
structures based on valence electrons and the
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Objective: The purpose of this experiment is for to visualize the number and types of bonds between atoms and how they are arranged. This will also allow for a better understand of isomers and conformers. **Discussion:** Organic chemistry focuses on the

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Explain how to use molecular shapes to predict molecular polarity. To predict molecule polarity from the shapes, you must first see if the molecule has exactly two atoms. If so, subtract the electronegatives to decide if it's polar. If the molecule has unshared electron pairs on the center atom (bent, triagonal pyramidal), the

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Name:_____ Introduction. Why should people care about the shapes of molecules? Consider that the properties of molecules, including their role in nature, depend primarily on molecular structures. Molecular shape determines a compound's boiling point, freezing point, viscosity, and the amount and type of its ...

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Chemical Bonds, Molecular Models, and Molecular Shapes. BACKGROUND

INFORMATION. The properties of chemical compounds are directly related to the ways in which atoms are bonded together to make molecules. Simple “ball-and-stick” models are commonly used to construct representations of some common molecules. These “ball-and-stick” models serve as a three-dimensional

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Chemical Bonds, Molecular Models, and Molecular Shapes

**Models Of Molecular Compounds Lab Molecular
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Construct 3-

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Objectives 1. Learn about the structures of covalent compounds and polyatomic ions. 2.

Draw Lewis structures based on valence electrons and the octet rule. 3. Construct

3-dimensional models of molecules and ions

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Alexa Butera Due: January 28, 2020 Period 8/9

Models of Covalent Compounds Purpose: The goal of this lab is to determine the polarity and shape of bonds in a molecule. Procedure: 1. Create the dot diagram corresponding to the molecule 2. With this, decide the shape of the molecule 3.

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To learn how to draw spatial representations and Newman projections, molecular models are useful. These give a simple representation of the

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Models Of Molecular Compounds Lab Chemistry 152L, Molecular Models Lab page 1 Revised 11/8/2009 Molecular Models Lab Objectives 1.

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Learn about the structures of covalent compounds and polyatomic ions. 2. Draw Lewis structures based on valence electrons and the octet rule. 3.

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Objective: The purpose of this experiment is for to visualize the number and types of bonds between atoms and how they are arranged. This will also allow for a better understand of isomers

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and conformers. Discussion: Organic chemistry focuses on the

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Objectives 1 Learn about the structures of covalent compounds and polyatomic ions 2

Draw Lewis structures based on valence electrons and the octet rule 3 Construct 3-

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