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In a chemical reaction, one or more reactants change into one or more products. Figure 11.1a shows the ingredients for making leavened bread—flour, salt, yeast, and water. Chemical reactions take place when the ingredients are mixed together and heated in the oven. Figure 11.1b shows the product—a loaf of bread.

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How can she conclude that the reaction started with three nitrogen atoms?

There are three nitrogen atoms in the products. The law of conservation of mass states that the amount of substance should be the same before and after a reaction.

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$$2\text{AgNO}_3 + 1\text{H}_2\text{S} \rightarrow 1\text{Ag}_2\text{S} + 2\text{HNO}_3$$
$$3\text{Zn}(\text{OH})_2 + 2\text{H}_3\text{PO}_4$$
$$1\text{Zn}_3(\text{PO}_4)_2 + 6\text{H}_2\text{O}$$
$$2\text{FeCl}_3 + 3\text{Ca}(\text{OH})_2 \rightarrow 2\text{Fe}(\text{OH})_3 + 3\text{CaCl}_2$$
$$1\text{CuS} + 1\text{O}_2$$
$$1\text{Cu} + 1\text{SO}_2$$
$$1\text{Fe}_2\text{O}_3 + 3\text{H}_2 \rightarrow 2\text{Fe} + 3\text{H}_2\text{O}$$

1. Propane gas reacts with oxygen to produce water vapor and carbon dioxide.

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