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call heat).

~~Unit 3 Lab: Icy Hot~~

1. Energy can be viewed as a substance-like quantity that can be stored in a physical system. 2. Energy can "flow" or be "transferred" from one system to another and so cause changes. 3. Energy maintains its identity after being transferred.

~~Chemistry – Unit 3: Energy and Kinetic Molecular Theory ...~~

Intro to Chemistry-Unit 3 test study guide. STUDY. PLAY. Energy. The ability to perform work or create a change in the environment. Law of Conservation of Energy. The energy in the Universe is constant.

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One important aspect of energy is that it is not created or destroyed, but it can be converted to one form from another. Thermal Energy.

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test, you should assemble and
review your lab notes, the 3
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the key points you should know.
States of Matter: From what we
have learned about states of matter
this unit, complete the following
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UNIT 3: Ionic Bonding Atoms can form ions because all atoms want full s and p subshells. A full s and p subshell (of the same energy level) is also called a “complete octet”, and is a very stable electron arrangement. Atoms therefore lose or gain their valence electrons to achieve the electron configuration of the nearest noble gas.

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The study is structured under a set of curriculum framing questions that reflect the inquiry nature of the discipline. Entry There are no prerequisites for entry to Units 1, 2

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and 3. Students must undertake Unit 3 prior to undertaking Unit 4.

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The notes in this article can be used to study smaller portions of the curriculum or to review for the final AP Chemistry exam. As of the 2019 updates, there are nine units that organize all the concepts in the course, so I've categorized these notes according to that framework.

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