

## Chapter 9 Stoichiometry Mixed Review

~~Stoichiometry Basic Introduction, Mole to Mole, Grams to Grams, Mole Ratio Practice Problems AP Chemistry: 9.1-9.3, 9.5, 7.14 Entropy and Gibbs Free Energy Significant Figures - A Fast Review! Introduction to Limiting Reactant and Excess Reactant~~

~~Calorimetry Problems, Thermochemistry Practice, Specific Heat Capacity, Enthalpy Fusion, Chemistry Converting Between Moles, Atoms, and Molecules Naming Ionic and Molecular Compounds | How to Pass Chemistry ATI TEAS 6 - MATH - Mixed Review - Part 11 - Money and Percent Word Problems ATI TEAS 6 - MATH - Mixed Review - Part 14 - Creating Equations AND A Bonus: Systems of Equations ATI TEAS 6 - MATH - Mixed Review - Part 2 - More Equations and Word Problems Writing Ionic Formulas: Introduction ATI TEAS 6 - MATH - Mixed Review - Part 13 - More Geometry and Distance Rate Time Word Problems~~

~~Mole Conversions Made Easy: How to Convert Between Grams and Moles Limiting Reactant Practice Problem (Advanced) How to Find Limiting Reactants | How to Pass Chemistry Stoichiometry Made Easy: The Magic Number Method How to Write Complete Ionic Equations and Net Ionic Equations Limiting Reactant Practice Problem TEAS Math Tutorial - Real World Percent Problems - Chapter 25 Limiting Reagent and Percent Yield~~

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the following equation:  $N_2(g) + 3H_2(g) \dots$

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CHAPTER 9 REVIEW. Stoichiometry. MIXED REVIEW. SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation:  $C_3H_4(g) + x O_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$  a. What is the value of the coefficient  $x$  in this equation? b. What is the molar mass of  $C_3H_4$ ? c. How many moles are in an 8.0 g sample of  $C_3H_4$ ? 2. a. What ...

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8.0 g sample of  $C_3H_4$ ? 2. a. What ...

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