

## Chapter 22 Hydrocarbons Study Guide For Content Mastery Answer

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Hydrocarbons part 1~~

Chapter 22: Part II - Hydrocarbons (Chem in 15 minutes or less)

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Chemistry Chapter 22 Sec 1 Organic compounds Part 1 Hydrocarbon

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this set (66) hydrocarbons ...

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22 CHAPTER Study Guide Key Concepts 22.1 Hydrocarbons •

Because carbon has four valence electrons, carbon atoms always

form four covalent bonds. • The carbon atoms in an alkane can be

arranged in a straight chain or in a chain that has branches. •

Molecules of hydrocarbons, such as alkanes, are nonpolar

molecules. 22.2 Unsaturated Hydrocarbons

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**A controlled process by which hydrocarbons are broken down or rearranged into smaller, more useful molecules natural gas This is typically made up of 80% methane, 10% ethane, 4% propane, and 2% butane**

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**718 Chapter 22 22CHAPTER Study Guide Key Concepts 22.1**

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### **Chapter 22 Hydrocarbons Study Guide Answers**

**Chapter 22: Substituted Hydrocarbons and Their Reactions. 784**

**Formic acid. BIGIdeaThe substitution of different functional groups for hydrogen atoms in hydrocarbons results in a diverse group of organic compounds. 22.1 Alkyl Halides and Aryl Halides. MAINIdeaA halogen atom can replace a hydrogen atom in some hydrocarbons.**

### **Chapter 22: Substituted Hydrocarbons and Their Reactions**

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### **Chapter 22 - Hydrocarbon Compounds - 22.1 Hydrocarbons ...**

**Chapter 22 Hydrocarbons Study Guide Answers Hydrocarbons are organic molecules that only contain two elements: carbon and hydrogen. Common hydrocarbons include methane, propane, and butane. Hydrocarbons are the primary constituents of fossil...**

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