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Chapter 19 Chapter 19: Acids and
Bases By: Loreley Estrada Stephanie~~

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acids and bases come in pairs. A

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Changes the color of an acid-base
indicator (acid, base, or both) Both.

Can be strong or weak electrolytes in
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citrus, pop, etc. Acid. This acid is associated with protein. amino acid. Used in fragrances and flavors.

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Acids. STUDY. PLAY. Acids. taste

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sour, change the color of an acid base indicator, can be strong or weak electrolytes in aqueous solution Ex: citrus. bases. taste bitter, feel slippery, will change the color of an acid base indicator, can be strong or weak electrolytes in aqueous solution. Ex: soap

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consider the equation for the ionization
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According to the Brønsted-Lowry definition, the equation is written this way. Agricultural Technician
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What are the conjugate acid-base pairs?

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What are the properties of acids and bases? acids taste sour, will change the color of an acid-base indicator, and can be strong or weak electrolytes in
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Chapter 19 Acids, Bases, and Salts. Acids and Bases: Basics. These are really just a specific type of chemical compound. No mysteries here. 1) They **MUST** be ionic compounds, and. **MUST** dissolve in water, that's the only

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way to be chemically active. We call this. disassociation.

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of an acid or base is dissolved in
solution - refers to the # of moles of
acid or base. Is a . concentrated,
weak. acid possible? Acid and Base

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Strength. Acetate is a stronger base than H₂O, LOWER on table, The stronger base “wins” the proton. ...

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conjugate acid-base pair amphoteric

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When ants sense danger to the ant colony, they emit a substance called formic acid that alerts the entire colony. Acids in rainwater hollow out enormous limestone caverns and destroy valuable buildings and statues. Acids flavor

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Chapter 19: Acids and Bases You're probably already aware from what you've learned in other classes that acids and bases are all over the place. From the lemon you put in iced tea to the Drano put down the drain to keep

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your hair from clogging the drain, acids and bases are an important part of our lives.

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OBJECTIVES: Define the properties of acids and bases. Acid-Base Theories

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Acids and Bases. SC6. Obtain,
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information about the properties that describe solutions and the nature of acids and bases. f. Use mathematics and computational thinking to compare, contrast, and evaluate the nature of acids and bases in terms of percent dissociation, hydronium ion concentration, and pH. g.

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