

Chapter 17 Acids Bases Worksheet

Chapter 17 (Additional Aspects of Aqueous Equilibria) - Part 1 Chapter 17 – Additional Aspects of Aqueous Equilibria: Part 1 of 21

17.1 Buffers

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for electron-withdrawing groups, as electronegativity increases, acidity increases. -as the number of oxygen atoms attached to Y (central atom) increases, acidity increases. Strong Acids. -acids that completely ionize in aqueous solution. 1. HCL (Hydrochloric acid) 2. HI (Hydroiodic acid) 3.

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