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+ H_2O . Ni^{2+} reacts
with concentrated
ammonia s.... Ni^{2+}
reacts with ethylene
diamine Ni^{2+}
 $\text{Cu}(\text{OH})_2 + 2 \text{NH}_3 + 2$
 NH_4^+ . $[\text{Ni}(\text{H}_2\text{O})_6]^{2+}$...
 $\text{Ni}^{2+} + 2 \text{NO}_3^-$...

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2. Nitrogen dioxide

(NO_2) has a bond

angle of 132° . What

does it imply

regarding its

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electronic and mo;
3. Suspension bridge
are known to be
normal around the
world especially in
developed countries.
These b; 4.

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aluminum is 2.70
g/mL. If you use 108
g of aluminum foil in
an experiment, what
is th; 3.If a can of dog
food contains 22.0 oz
(ounces) of dog food,

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how many grams of
dog food are in the
ca; 4. Milk has a
density of 1.04 g/mL.
What is the mass in
grams of 1.00 quart
of milk? a. 984 g b.
0.98; 5.

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experiment. $\text{Ni}(\text{NO}_3)_2$
+ H_2O . Ni^{2+} reacts
with concentrated
ammonia s.... Ni^{2+}
reacts with ethylene
diamine Ni^{2+}
 $\text{Cu}(\text{OH})_2 + 2 \text{NH}_3 + 2$

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$\text{Ni}^{2+} + 2 \text{NO}_3^-$...

hexaaquanickel (II)...

green coloured....

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reaction mixture and
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regarding its
electronic and mo;
3. Suspension bridge
are known to be
normal around the
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an experiment, what
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